

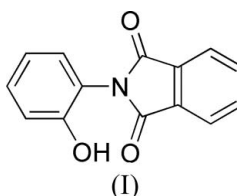
## 2-(2-Hydroxyphenyl)isoindoline-1,3-dione

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## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.048  
 $wR$  factor = 0.146  
Data-to-parameter ratio = 13.6For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The crystal structure of the title compound,  $\text{C}_{14}\text{H}_9\text{NO}_3$ , is stabilized by a weak  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond. The dihedral angle between the phthalimide and hydroxyphenyl planes is  $71.8(2)^\circ$ .

## Comment

Phthalimides and *N*-substituted phthalimides are an important class of compounds because of their interesting biological activities (Lima *et al.*, 2002). We report here the structure of the title compound, (I) (Fig. 1).The phthalimide and hydroxyphenyl units are essentially planar, with mean deviations of  $0.007(4)$  and  $0.006(2)$  Å, respectively. The dihedral angle between these planes is  $71.8(2)^\circ$ . The geometry of the molecule is close to that of the related compound 2-(4-hydroxyphenyl)isoindoline-1,3-dione (Liang & Li, 2006). The crystal structure is stabilized by a weak  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond (Table 1).

## Experimental

A mixture of phthalimide (0.1 mol), 2-aminophenol (0.1 mol) and triethylamine (0.01 ml) in xylene (100 ml) was refluxed for 3 h. After cooling, filtration and drying, the title compound was obtained. Yellow single crystals suitable for X-ray analysis were obtained by slow evaporation of an acetone solution for 8 d at room temperature.

## Crystal data

$\text{C}_{14}\text{H}_9\text{NO}_3$	$Z = 4$
$M_r = 239.22$	$D_x = 1.465$ Mg m $^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 11.584(2)$ Å	$\mu = 0.11$ mm $^{-1}$
$b = 13.704(3)$ Å	$T = 294(2)$ K
$c = 7.0359(13)$ Å	Block, yellow
$\beta = 103.877(3)^\circ$	$0.30 \times 0.24 \times 0.20$ mm
$V = 1084.4(4)$ Å $^3$	

## Data collection

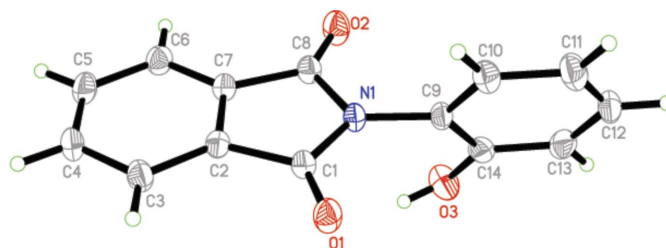
Bruker SMART CCD area-detector diffractometer	6010 measured reflections
$\varphi$ and $\omega$ scans	2225 independent reflections
Absorption correction: multi-scan (SADABS; Bruker, 1997)	1674 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.969$ , $T_{\max} = 0.979$	$R_{\text{int}} = 0.029$
	$\theta_{\text{max}} = 26.4^\circ$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.146$   
 $S = 1.05$   
 2225 reflections  
 164 parameters  
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0748P)^2 + 0.4133P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.59 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.26 \text{ e } \text{Å}^{-3}$



**Figure 1**  
 The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level.

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O3-H3\cdots O1^i$	0.82	2.00	2.798 (2)	163

Symmetry code: (i)  $x, -y + \frac{3}{2}, z - \frac{1}{2}$ .

H atoms were positioned geometrically ( $C-H = 0.93 \text{ Å}$  and  $O-H = 0.82 \text{ Å}$ ) and refined as riding, with  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$  and  $1.5U_{\text{eq}}(O)$ .

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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